

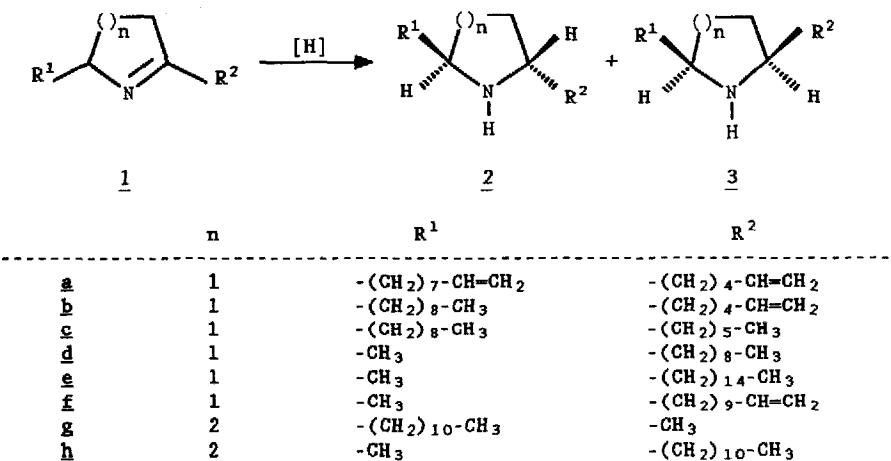
REDUCTION OF 2,5-DIALKYL PYRROLINES . A KEY STEP IN A SYNTHESIS OF NATURAL INSECTICIDES

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**Abstract :** The reduction of 2,5-dialkylpyrrolines can be run with a correct selectivity.

We have recently shown that the venom of the ant *Monomorium Minutum* is composed of five alkaloids : two pyrrolines 1a and 1b and three trans pyrrolidines 2a,b and c<sup>1,2</sup>.



2,5-Dialkylpyrrolidines can be obtained by the aminoreduction of 1,4-diketones as 1/1 cis and trans mixture<sup>3</sup>, like by the bis alkylation  $\alpha$ -to the amino group of formamidines<sup>4</sup>. The stereospecific mercurycyclization<sup>5</sup> of unsaturated amines or other nitrogen derivatives, leads to a stereoselectivity of the reduction by borohydride, which removes the mercury atom. Reduction of pyrroles<sup>6</sup>,  $\Delta$ -3 pyrrolines or cyclic iminiums<sup>7</sup> is more cis selective, but excluded for unsaturated substituents.

Only one stereospecific synthesis of trans 2,5-dialkylpyrrolidines has been described, using a bis nitrone-alkene cycloaddition reaction<sup>8</sup>, which is inefficient with  $\omega$ -dienes. Recently, H. Yamamoto<sup>9</sup> reported a highly stereoselective reduction of 2,6-dialkylpiperidine 1g by DIBAH leading either to the cis piperidine 3g or to the trans piperidine 2g using [LiAlH<sub>4</sub>-Me<sub>3</sub>Al] and we checked that the same stereoselectivity is observed for the reduction of the piperideine 1g when a long chain is present instead of a methyl substituent.

Three models of natural pyrrolines were used to run the reduction with hydride reagents : 1d and 1c with a long alkyl chain and 1f with an alkenyl substituent.

Dedicated to Professor A.R. Katritzky, FRS, on the occasion of his 60<sup>th</sup> birthday.

**Table 1 : Stereoselectivity in the reduction of the imine 1 with hydride reagents<sup>a</sup>**

Run	Compound	Hydride reagent	Solvent	Ratio <sup>b</sup> (2:3)
1	<u>1h</u>	LiAlH <sub>4</sub> (7 eq)-Me <sub>3</sub> Al(7 eq)	THF	95:5
2,3,4	<u>1d,e,f</u>	DIBAH(4 eq)	CH <sub>2</sub> Cl <sub>2</sub>	0:100
5	<u>1f</u>	DIBAH(4 eq)-Ti(O <i>i</i> Pr <sub>2</sub> ) <sub>4</sub> (7 eq)	CH <sub>2</sub> Cl <sub>2</sub>	0:100
6	<u>1d</u>	LiAlH <sub>4</sub> (25 eq)	Et <sub>2</sub> O	20:80
7	<u>1e</u>	LiAlH <sub>4</sub> (7 eq)-Ti(O <i>i</i> Pr <sub>2</sub> ) <sub>4</sub> (7 eq)	THF	20:80
8	<u>1e</u>	LiAlH <sub>4</sub> (7 eq)-TiCl <sub>4</sub> (7 eq)	THF	30:70
9	<u>1f</u>	LiAlH <sub>4</sub> (7 eq)-Me <sub>3</sub> Al(7 eq)	Et <sub>2</sub> O	40:60
10	<u>1f</u>	LiAlH <sub>4</sub> (7 eq)-Me <sub>3</sub> Al(7 eq)	THF	10:90
11	<u>1e</u>	LiAlH <sub>4</sub> (7 eq)-Ni(acac) <sub>2</sub> (7 eq)	Et <sub>2</sub> O	30:70
12	<u>1f</u>	NaBH <sub>3</sub> CN(3 eq)-HCl	MeOH	50:50
13	<u>1e</u>	NaBH <sub>3</sub> CN(3 eq)	AcOH	35:65
14,15	<u>1e,f</u>	NaBH <sub>4</sub> (3 eq) or (10 eq)	EtOH or Et <sub>2</sub> O	50:50
16,17	<u>1e</u>	NaBH <sub>4</sub> (10 eq)-TiCl <sub>4</sub> (6 eq)	CH <sub>2</sub> Cl <sub>2</sub> or Et <sub>2</sub> O	40:60
18	<u>1e</u>	NaBH <sub>4</sub> (3 eq)-NiCl <sub>2</sub> (5 eq)	EtOH	40:60
19	<u>1f</u>	NaBH <sub>4</sub> (3 eq)-PdCl <sub>2</sub> (1.3 eq)	EtOH	0:100
20	<u>1e</u>	NaBH <sub>4</sub> (3 eq)	HCOOH	50:50
21	<u>1e</u>	NaBH <sub>4</sub> (3 eq)	AcOH	70:30
22	<u>1e</u>	NaBH <sub>4</sub> (3 eq)	iPrCOOH	55:45
23	<u>1e</u>	NaBH <sub>3</sub> (OAc) or NaBH(OAc) <sub>3</sub> (3 eq)	THF	35:65
24	<u>1e</u>	NaBH <sub>4</sub> (3 eq)-H <sub>3</sub> BO <sub>3</sub> (3 eq)	CH <sub>2</sub> Cl <sub>2</sub>	50:50
25	<u>1e</u>	NaBH <sub>4</sub> (3 eq)-Silicagel	CH <sub>2</sub> Cl <sub>2</sub>	65:35

<sup>a</sup>-Unless specified, the experiment afforded 2 and 3 in >93% yield<sup>b</sup>-The ratio was determined by <sup>13</sup>C n.m.r.

DIBAH always leads stereospecifically to the cis isomer **3** with or without Lewis'acid (entries 2-5) like the piperideine reduction of compound **1g**. Besides, LiAlH<sub>4</sub> alone or in presence of TiCl<sub>4</sub>, Ti(O-iPr<sub>2</sub>)<sub>4</sub> or Me<sub>3</sub>Al exclusively leads to the cis pyrrolidine; these results are in complete opposition to those obtained during the reduction of piperideine **1g** (entries 6-11). Meanwhile, NaBH<sub>4</sub> gives an equal mixture of **2** and **3**, but with NiCl<sub>2</sub> or TiCl<sub>4</sub> the major component obtained is always the cis isomer (entries 16-18). However, the cis pyrrolidine **3** can only be isolated in the presence of PdCl<sub>2</sub>, but here, the C=C double bond is unfortunately also reduced. To stereoselectively prepare the trans isomer **2**, NaBH<sub>4</sub> reduction was carried out in acidic medium : acetic acid gives the best ratio (70:30). This reduction study will permit a stereoselective or specific access to natural pyrrolidines **2a-c** which exhibit a very broad and significant insecticidal activity<sup>1b</sup>.

## REFERENCES

1. a)-Bacos, D., Basselier, J.J., Célérier, J.P., Lange, C., Marx, E., Lhommet, G., Escoubas, P., Lemaire, M., Clément, J.L., *Tetrahedron Lett.*, 1988, **29**, 3061.  
b)-Cassier, P., Clément, J.L., Basselier, J.J., Lange, C., Célérier, J.P., Lhommet, G., French Patent N° 84/06980; *Chem. Abstr.*, 1986, **104**, 20233
2. Will be presented in *Biomed. Environ. Mass Spectrom.*
3. Jones, T.H., Blum, M.S., Howard, R.W., Mc Daniel, C.A., Fales, H.M., Dubois, M.B., Torres, J., *J. Chem. Ecol.*, 1982, **8**, 285.
4. Meyers, A.I., Edwards, P.D., Bailey, T.R., Jagdmann, G.E., *J. Org. Chem.*, 1985, **50**, 1019.
5. a)-Harding, K.F., Burks, S.R., *J. Org. Chem.*, 1981, **46**, 3920.  
b)-Barlett, P.A., Olefin Cyclization Process that Form Carbon-Heteroatom Bonds, *Asymmetric Synthesis*, 1984, vol III, p. 342, Ed. Morrison, J.D., Academic Press
6. a)-Stillman, N., Osgood, E., U.S. Patent N° 3 864 361 (1975).  
b)-Benezza, L., German Patent N° 2 407 096 (1974).  
c)-Benezza, L., De Pablo, P., Osgood, E., German Patent N° 2 344 509 (1974).
7. a)-Evans, G., *J. Am. Chem. Soc.*, 1951, **73**, 5230.  
b)-Mc Donald, T.L., *J. Org. Chem.*, 1980, **45**, 193.  
c)-Shiosaki, K., Rapoport, H., *J. Org. Chem.*, 1985, **50**, 1229.
8. Tufariello, J.J., Puglis, J.M., *Tetrahedron Lett.*, 1986, **27**, 1489.
9. a)-Matsumura, Y., Maruoka, K., Yamamoto, H., *Tetrahedron Lett.*, 1982, **23**, 1929.  
b)-Maruoka, K., Miyazaki, T., Ando, M., Matsumura, Y., Sakane, S., Hattori, K., Yamamoto, H., *J. Am. Chem. Soc.*, 1983, **105**, 2831.

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